Accurate assessment of the Hirakud smelter aluminium reduction cell thermal balance using only temperature measurements

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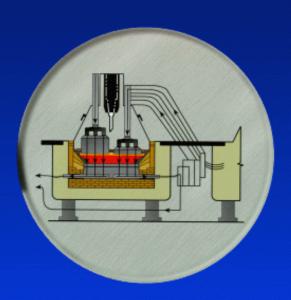


Plan of the Presentation

- Introduction
- The cell heat balance
- Cell heat losses measurement methods
- Fundamental heat flux equations
- Measurement campaign
- Analysis of the results
- Conclusions







The power consumption of the Hall-Héroult cell being one of the major operating costs, the aluminium industry is constantly trying to reduce the specific power consumption of smelters expressed in kWh/kg of aluminium produced.

Today, best results are:

12.9 - 13.0 kWh/kg for high amperage PBF cells

14.0 - 14.5 kWh/kg for best VSS cells

Older smelters still operating at 17 - 18 kWh/kg are feeling an increasing pressure from their more efficient competitors. They have essentially two options:

- 1) Retrofit their cell design in order to improve their power consumption and hence reduce their production costs
- 2) Be run out of business

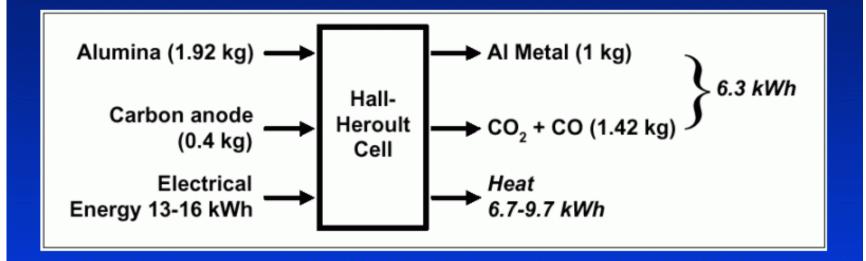


Introduction



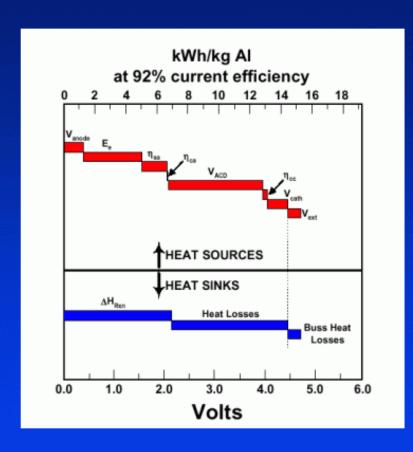
The first priorities of the Hirakud smelter cell retrofit program has been to measure the cell heat balance and to develop reliable mathematical models.





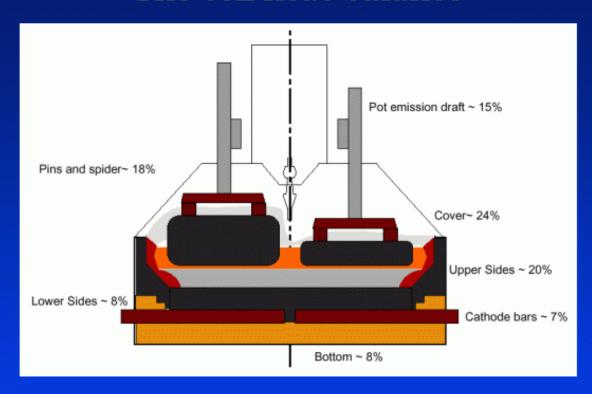
The concept of the cell heat balance is quite simple: of the total electrical power fed to the cell, less than half is actually used to produce aluminium, the remaining part must be dissipated as heat losses in order for it to maintain its thermal equilibrium.





Experimentally, a cell voltage break down is required in order to calculate the cell internal heat: the heat that the cell needs to dissipate to maintain its thermal equilibrium.





This can be experimentally confirmed by directly measuring the cell heat losses.



Measurement of the complete heat balance requires that all heat losses be measured such that the sum of the measurements is equal to what is called the theoretical heat loss. The theoretical heat loss is calculated from the voltages and line current. The heat balance closure, then, is defined as the % of the theoretical heat that is actually measured:

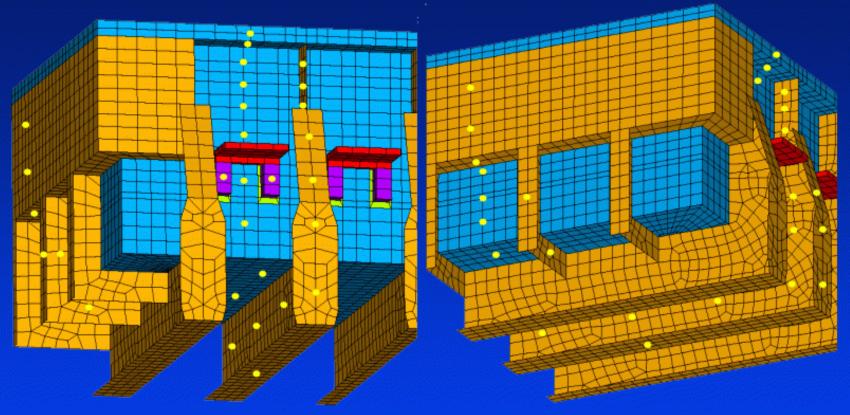
% Closure =
$$\frac{Q_{measured}}{\left(V_{pot} - E_{\Delta H} - V_{ext}\right) \bullet I} (100\%)$$

Thorough and careful measurements usually close the heat balance between 93% and 105% of the theoretical heat loss.

If the cell heat losses correspond to the calculated cell internal heat, those measurements can be used with confidence in order to calibrate the mathematical models of the cell.

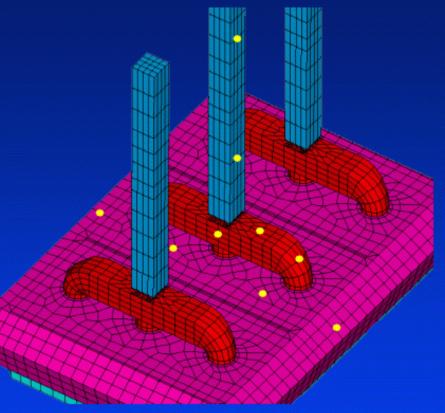






In order to calculate the global cell heat losses, approximately 200 surfaces must be established around the cell.





The area of each of those surfaces must be calculated in order to be able to, in turn, calculate the heat dissipated by each of them.

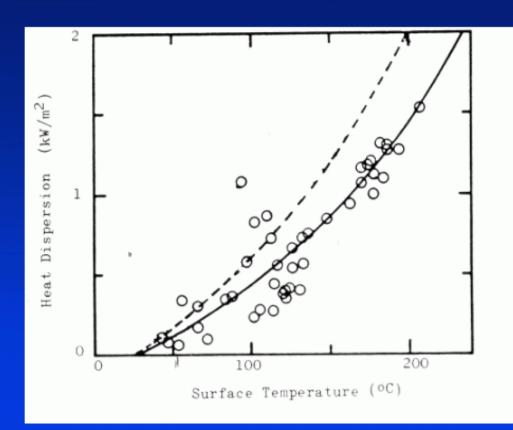


Heat Balance Results				
date: 12-Aug-03	Cell:	*VAW* 300		
Cathode Heat Losses	W / m2	kW	*	
Shell side wall above bath level Shell side wall opposite to bath Shell side wall opposite to setal Shell side wall opposite to block above bar Shell side wall opposite to block between bars Collector bars to air Collector bars to flexible Shell side wall opposite to brick Shell floor close to corner Shell floor quarter point region Shell floor quarter point region Cradle above bath level Cradle opposite to bath Cradle opposite to bath Cradle opposite to block above bar Cradle opposite to block above bar Cradle opposite to block corner Cradle opposite to brick Cradle opposite to brick Cradle opposite to brick Cradle opposite to brick Shell floor close to corner Cradle below floor close to corner Cradle below floor close to corner Cradle below floor cuarter point region Cradle below floor cuarter point region Shell end wall opposite to block above bar Shell end wall opposite to block below top of bar Shell end wall opposite to block below top of bar Shell end wall opposite to block below top of bar Shell end wall opposite to brick Shell horizontal strip in the ends Shell horizontal striffeners in the ends	2002 5616 7617 6041 1505 3035 990 504 499 498 882 1906 2360 854 355 130 101 102 1503 3050 4034 2991 4991 4991 4991 4991 837	11.53 32.35 43.87 48.71 6.50 17.48 57.42 11.40 12.65 10.42 8.31 6.04 13.04 16.14 8.18 2.43 1.78 1.53 2.75 2.80 2.77 2.80 7.02 10.20 1.49 27.59 5.65 380.56	1.83 5.14 6.97 7.74 1.03 2.78 9.12 1.81 1.65 1.32 0.96 2.07 2.56 1.30 0.39 0.24 0.44 0.44 0.45 1.18 1.62 0.24 4.38 0.90 0.24	
Total for the cathode part Anode Heat Losses		380.56	60.43	
Crust in side channels Crust above anodes Crust in center channel Studs Yoke Aluminum rod Total for the anode part	1713 1798 1740 4002 3682 818	21.65 81.80 3.58 27.16 84.83 30.14 249.2	3.44 12.99 0.57 4.31 13.47 4.79	
Total for the cell		629.7	100.00	

For each surfaces, the average heat flux is multiply by the surface area in order to obtain the surface heat loss.

The summation of all the surfaces heat losses gives the total cell heat loss.





In the early cell heat balance measurement campaigns, the equations that describe the physics of the two heat transfer mechanisms (natural convection and radiation) were used to correlate the different cell surface temperatures to the heat fluxes in order to reduce it to the measurement of the cell surface temperatures only.

This approach turned out not to produce very accurate results.









Bruggeman[1], clearly expressed the current conventional wisdom of the industry by specifying that heat flux transducer must be used to carry out cell heat loss measurement campaigns.





He even pointed out that: "Haupin developed a heat flux transducer especially for pot measurements". Apart from Alcoa, most of the rest of the industry is rather using commercially available heat flux transducers.



Unfortunately, commercially available heat flux transducers are fairly expensive and relatively fragile.

They are also characterized by a fairly long response time of around 10 minutes. This means that cell heat balance measurement campaigns using heat flux transducers are fairly long and expensive to carry out.

For that reason, the challenge posed to the authors was to find an accurate way to assess the Hirakud cell thermal balance using only temperature measurements.



$$q_{tot}(W/m^{2}) = q_{c} + q_{r}$$

$$q_{c}(W/m^{2}) = Nu \cdot \frac{k}{L} \cdot (T_{S} - T_{A})$$

$$q_{r}(W/m^{2}) = \varepsilon \sigma \left((T_{S} + 273)^{4} - (T_{O} + 273)^{4} \right)$$

The equations that describe the physics of natural convection and radiation heat transfer mechanisms are well known.

For vertical surfaces, we have:

$$Nu = 0.59 Ra^{\frac{1}{4}}$$
, for $10^4 \le Ra \le 10^9$

$$Nu = 0.105Ra^{\frac{1}{3}}$$
, for $10^9 \le Ra \le 10^{12}$

For horizontal surfaces facing up we have:

$$Nu = 0.54 Ra^{\frac{1}{4}}$$
, for $10^5 \le Ra \le 2 \times 10^7$

$$Nu = 0.141Ra^{1/3}$$
, for $10^7 \le Ra \le 10^{11}$

And finally, for horizontal surfaces facing down we have:

$$Nu = 0.27 Ra^{\frac{1}{4}}$$
, for $3 \times 10^5 \le Ra \le 3 \times 10^{10}$

The general form of those heat transfer equations have been published multiple times[2,3].

Notice that the natural convection heat loss equations are semi-empirical.



Where:

$$Ra = \frac{g\beta L^{3}(T_{S} - T_{A})}{v^{2}} \cdot \Pr$$

$$k = 2.014E - 15 \times T_f^4 + 1.68E - 11 \times T_f^3 - 4.118E - 8 \times T_f^2 + 8.051E - 5 \times T_f + 0.02407$$

$$\upsilon \!=\! 1.438E \!-\! 17 \times T_f^4 - 3.25E \!-\! 14 \times T_f^3 + 9.095E \!-\! 11 \times T_f^2 + 8.977E \!-\! 8 \times T_f + 1.32E \!-\! 5$$

$$\Pr = 1.937E - 13 \times T_f^4 - 6.581E - 10 \times T_f^3 + 7.349E - 7 \times T_f^2 - 2.788E - 4 \times T_f + 0.714$$

$$T_f = \frac{\left(T_S + T_A\right)}{2}$$

 $q_{tot}(W/m^2) = F_q(T_S, T_A, T_O, \varepsilon, L, SO)$

Where:

T_S (°C) is the measured surface temperature

T_A (°C) is the measured air temperature close to the surface

T_O (°C) is the measured facing radiative background temperature

ε is the surface emissivity

L (m) is the surface typical length

SO is the surface orientation (V, OH or OD)

Measurement campaign

In order to calculate the global cell heat losses, the area of each representative surfaces must be calculated in order to be able to, in turn, calculate the heat dissipated by each of them.

$$Q_i(W) = A_i \cdot F_q(T_S, T_A, T_O, \varepsilon, L, SO)_i$$

$$Q_{cell}(kW) = \sum_{1}^{N} Q_i / 1000$$

It is quite easy to evaluate ahead of time A, E, L and SO for each surface. For a given cell design, once established, the value of those items will not change.



Measurement campaign

This leaves only three temperatures to be measured per surface T_S , T_A and T_O during the measurement campaign (instead of only T_S for "early" measurement campaign).

In comparison, in a "standard" measurement campaign using heat flux transducers, only \mathbf{q}_{tot} is measured for each surface.

Yet, measuring $T_{\rm S}$ and $T_{\rm O}$ with a small hand held pyrometer and $T_{\rm A}$ with a thermocouple and a small hand held multimeter is far less cumbersome and requires far less time than using slow to respond heat flux transducers.



Analysis of the results

Analysis of the initial results revealed that we were overestimating the heat flux of some very hot surfaces for which the radiation term became very large.

Of course, we knew that the radiative exchanges around a cell are very complex and that considering that each surface is only seeing one background radiative temperature could well turn out to be an unrealistic over-simplification.



Analysis of the results

This led us to consider that those surfaces are seeing two background radiative objects that are not at the same temperature. Of course, this in turn introduce the need to calculate the view factors for each object:

$$q_r(W/m^2) = \varepsilon\sigma \Big[F\Big((T_S + 273)^4 - (T_O + 273)^4 \Big) + (1 - F)\Big((T_S + 273)^4 - (T_O' + 273)^4 \Big) \Big]$$



Analysis of the results

Heat Balance Results				
date: 5-Nov-03	Cell:	Cell 265		
Cathode Heat Losses	W / m2	kW	*	
Shell side coverplate Shell side spacer between boxes Shell side bottom box Collector bars to air Collector bars to flexible Shell side wall collector bar level Shell side wall insulation level Shell side vertical boxes Shell side floor perimeter section Shell side floor center section Shell end coverplate Shell end spacer between boxes Shell end wall collector bar level Shell end wall insulation level	821 1796 3105 1337 625 981 918 1172 1884 809 1789 3065 523 969	2.26 3.59 6.41 7.96 2.40 0.57 1.80 1.97 9.56 8.14 2.21 3.55 6.29 0.81	2.00 3.18 5.69 7.06 2.13 0.50 1.60 1.75 8.48 7.22 1.96 3.15 5.58 0.72	
Shell end vertical boxes Shell end floor perimeter section Total for the cathode part	1150 1165	2.16 6.34 67.82	1.92 5.62 60.17	
Anode Heat Losses				
Crust First side channel Second side channel Third side channel Forth side channel Above forth side channel Anode top Studs	306 4002 2037 1206 739 554 552	2.85 12.11 6.16 3.65 2.24 1.70 4.13 12.06	2.53 10.74 5.47 3.24 1.98 1.51 3.67	
Total for the anode part		44.9	39.83	
Total for the cell		112.7	100.00	
Cell internal heat Blitz closing		117.8 95.71%		

At the second trial, the percentage of closure was in the acceptable rage.



Conclusions

- The heat balance of the Hirakud smelter aluminium reduction cell has been accurately assessed using only a surface thermocouple and a pyrometer.
- No loss in accuracy was detected when using this approach as the cell heat balance could be closed within 5%, the level of accuracy typical of cell heat balance assessment using heat flux meter.
- Although at least three temperature measurements are required to estimate a single heat flux, it takes a lot less time to make those temperature measurements using a pyrometer than directly measuring the heat flux using an expensive heat flux meter.

